

## Research Article

Synthesis of  $\alpha, \beta, \gamma$ -triazole- $\delta$ -thiol derivative

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## Abstract

Several new quinoline- $\xi$ -carboxylic acid derivatives were synthesized via pfitzinger reaction and tested for their anti-inflammatory and ulcerogenic effects compared to NSAIDs. Replacement of the ulcerogenic carboxylic (COOH) functional group with its less acidic bioisostere  $\alpha, \beta, \gamma$ -triazole ring is expected to lower the ulcerogenic potentials of quinoline- $\xi$ -carboxylic acid derivatives as well as NSAIDs.

**Key Words:**  $\alpha, \beta, \gamma$ -triazole- $\delta$ -thiol, Quinoline, Anti-inflammatory

## Introduction

Quinoline is nitrogen containing heterocyclic aromatic compound. Quinoline derivatives have a wide therapeutic applications such as anti-malarial<sup>[1]</sup>, analgesic<sup>[2]</sup>, anticancer<sup>[3]</sup>, anti-inflammatory activities<sup>[4]</sup> and in treating of alzheimer's disease<sup>[5]</sup>.

There are so many reported procedures for the synthesis of quinoline ring<sup>[6]</sup>, the main core of quinoline ring has been synthesized via well-known named chemical reactions as Skraup<sup>[7]</sup>, Doebner-Von Miller<sup>[8]</sup>, Friedländer<sup>[9]</sup>, Pfitzinger<sup>[10]</sup>, Conrad-Limpach<sup>[11]</sup> and Combes syntheses<sup>[12]</sup>. Nowadays scientists are focusing on introduction of novel and safe therapeutic drugs with new scaffolds of clinical importance.

Triazole ring has been incorporated into a variety of pharmacologically active drugs, it can be synthesized via numerous chemical reactions as Pellizari Reaction<sup>[13]</sup> and Einhorn-Brunner synthesis<sup>[14]</sup>. Triazoles have many therapeutic applications such as anti-microbial<sup>[15]</sup>, anti-inflammatory<sup>[16]</sup>, anticancer<sup>[17]</sup>, anticonvulsant<sup>[18]</sup> and anti-malarial activities<sup>[19]</sup>.

## Results and discussion

## Chemistry

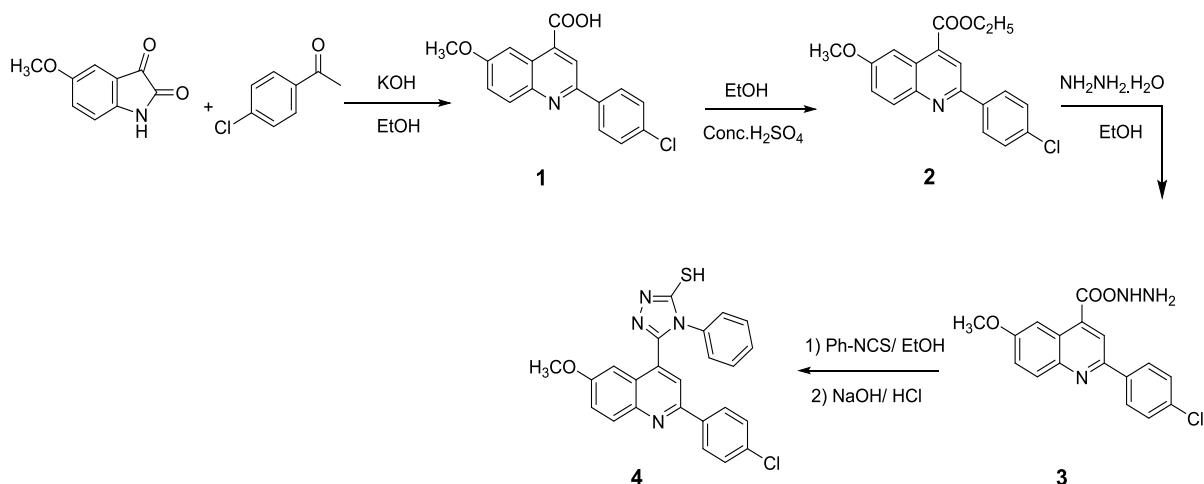
The synthesis of the target compound  $\alpha$ -( $\beta$ -( $\xi$ -chlorophenyl)- $\gamma$ -methoxyquinolin- $\xi$ -yl)- $\xi$ -phenyl- $\xi$ -H- $\alpha, \beta, \gamma$ -triazole- $\delta$ -thiol  $\xi$  is illustrated in Scheme 1.

Synthesis of  $\beta$ -(chlorophenyl)quinoline- $\xi$ -carboxylic acid  $\beta$  was carried out according to the reported method<sup>[20]</sup> by heating at reflux of  $\alpha$ -methoxyisatin with *p*-chloroacetophenone in aqueous ethanol affording compound  $\beta$  in a good yield. Treatment of the acid  $\beta$  with absolute ethanol in presence of conc. H<sub>2</sub>SO<sub>4</sub> affording the corresponding esters  $\gamma$ . Refluxing of ester with hydrazine monohydrate affording the carbohydrazide derivative  $\delta$ . Structure of the formed compounds have been confirmed by their reported melting points.

Heating at reflux of the carbohydrazide  $\delta$  and phenyl isothiocyanate in ethanol followed by the addition of  $\beta$ N NaOH and acidification with conc. HCl affording the final  $\alpha, \beta, \gamma$ -triazole- $\delta$ -thiol derivative  $\xi$ .

The purity of compound  $\xi$  has been checked by TLC and the structure was confirmed by <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectra.

**Scheme 1:** Synthesis of  $\alpha$ -( $\gamma$ -( $\delta$ -chlorophenyl)- $\beta$ -methoxyquinolin- $\epsilon$ -yl)- $\delta$ -phenyl- $\epsilon$ -H- $\gamma$ , $\delta$ -triazole- $\zeta$ -thiol ( $\xi$ ).



The  $^1\text{H-NMR}$  spectrum for the  $\gamma$ , $\delta$ , $\epsilon$ -triazole- $\zeta$ -thiol derivative  $\xi$  is characterized by the appearance of  $\text{SH}$  proton at 11.62 ppm and the protons of the methoxy group ( $\text{OCH}_3$ ) at 3.94 ppm.

In  $^{13}\text{C-NMR}$  spectrum, the methoxy carbon ( $\text{OCH}_3$ ) appeared at 50.81 ppm and the  $\gamma$ , $\delta$ , $\epsilon$ -triazole carbon attached to the thiol group ( $\text{C-SH}$ ) appeared at 109.20 ppm and this is may be attributed to the high electronegativity of sulfur atom.

## Experimental

### General

Melting points were determined on Stuart electro-thermal melting point apparatus and were uncorrected.  $^1\text{H-NMR}$  and  $^{13}\text{C-NMR}$  spectra were recorded in Umm-al-Qura University, Saudi Arabia using TMS as reference standard and  $\text{CDCl}_3$  as solvent. Chemical shifts are expressed in parts per million (ppm) and coupling constants ( $J$ ) are expressed in Hertz.

### Synthesis of compounds 1-3.

$\gamma$ -( $\delta$ -Chlorophenyl)- $\beta$ -methoxyquinoline- $\epsilon$ -carboxylic acid, ethyl  $\gamma$ -( $\delta$ -chlorophenyl)- $\beta$ -methoxyquinoline- $\epsilon$ -carboxylate and  $\gamma$ -( $\delta$ -chlorophenyl)- $\beta$ -methoxyquinoline- $\epsilon$ -

carbohydrazide were prepared according to the reported procedures. [21]

### Synthesis of $\alpha$ -( $\gamma$ -( $\delta$ -chlorophenyl)- $\beta$ -methoxyquinolin- $\epsilon$ -yl)- $\delta$ -phenyl- $\epsilon$ -H- $\gamma$ , $\delta$ , $\epsilon$ -triazole- $\zeta$ -thiol ( $\xi$ ).

Equimolar quantities of the carbohydrazide **3** (100 mmol) and phenyl isothiocyanate (100 mmol) in 120 ml of absolute ethanol were heated at reflux for  $\xi$  h. The solvent was evaporated under reduced pressure. Then 100 ml of 2N NaOH solution was added and the obtained solution was refluxed for 2 h. The reaction mixture was cooled and acidified to pH 2 with concentrated HCl. The solid that precipitated was filtered off, washed with water, and recrystallized from ethanol. Compound  $\xi$  was obtained as a white solid, (in 79% yield), m.p. 238-240°C;  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 11.63 (s, 1H,  $\text{SH}$ ), 8.27-8.10 (m, 1H, Ar-H), 7.77-7.72 (m, 2H, Ar-H), 7.07-7.00 (m, 2H, Ar-H), 6.88-6.82 (m,  $\epsilon$ H, Ar-H), 6.3-6.22 (m, 1H, Ar-H), 6.28 (s, 2H, Ar-H), 6.21 (d,  $J$  = 7.9 Hz, 1H, Ar-H), 3.90 (s, 3H,  $-\text{OCH}_3$ );  $^{13}\text{C NMR}$  (120 MHz,  $\text{CDCl}_3$ )  $\delta$  (ppm) 109.26, 143.72, 133.78, 133.09, 130.70, 130.24, 129.78, 129.12, 128.42, 128.37,

128.13, 127.71, 127.40, 127.90, 127.07,  
123.89, 120.04, 103.06, 102.47, 00.82.

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